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1

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Pure Substances and

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\u0026 Compounds,

Classification of Matter,

Chemistry Examples,

3.2.1/3.2.2 Describe the

differences between

elements, compounds

and mixtures. ~~States of~~

~~Matter : Solid Liquid~~

~~Gas 10 Amazing~~

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~~Experiments with Water~~

Changes in the State of
Matter - Science

Experiments - Solid to
Liquid, Liquid to Gas,

Gas to Liquid Changes
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A change in physical form of a substance or physical properties of a substance like color, melting point, state, size, shape, and arrangement of molecules is known as

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physical change. A change in chemical properties of a substance like composition, reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be formed.

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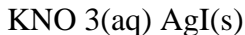
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CHAPTER 9

SOLUTIONS

MANUAL 35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide.

chemical equation:



complete ionic equation:

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$K(aq) + I_2(aq) \rightarrow 2Ag(aq) + NO_3^-(aq) + K^+(aq) + NO_3^-(aq) + AgI(s)$

Chemical

Reactions

Reactions

Solution. Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written: H_2O

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\left(l \right)

\rightarrow \text{H}_2\text{O}

\left(g \right)

Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...

240 Chemistry: Matter

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CHAPTER 12

SOLUTIONS

MANUAL 30. Compare

and contrast sublimation

and evaporation. In both

processes, the

substances become a

vapor. During

sublimation, the

substance goes from the

solid phase directly to

the vapor phase. During

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evaporation, particles in
a liquid gain enough

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