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CHAPTER 10 REVIEW States of Matter SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two opposing changes occur at equal rates. (b) there are no more phase changes.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting

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Holt: Modern Chemistry Mickey Sarquis, Jerry L. Sarquis. Chapter 10 States of Matter. Educators. Chapter Questions. Problem 1 What idea is the kinetic-molecular theory based on? Check back soon! Problem 2 What is an ideal gas? Check back soon! Problem 3 ...

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Name Date Class CHAPTER 10 REVIEW States of Matter SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ideal gas

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