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Solution:  $p_{abs} = p_{atm} + p_{gage} = 101:3kPa + 20 \text{ in. H}_2\text{O} = 1248:8Pa$   $1:0 \text{ in. H}_2\text{O} = 110:26kPa$  Ideal gas law:  $\rho = \frac{p}{RT} = \frac{110:26kPa}{1 \text{ kgK} \cdot 2077J}$   $1:293:2K = 1000Pa$   $1kPa = 1000Pa$   $\rho = 0:181 \text{ kg/m}^3$  b.) Situation: A sphere of 93 mm diameter contains an ideal gas.  $T = 20 \text{ C} = 293:2K$  Find: Calculate the density of argon at a vacuum pressure of 8.8 psi. Properties:

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2.5 Situation: Specific gravity (select all that apply) a. can have units of N/m<sup>3</sup> b. is dimensionless c. increases with temperature d. decreases with temperature SOLUTION Correct answers are b and d.

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